

absorbancia

$$\lambda = 410 \text{ nm}$$

$$\epsilon = 25000 \frac{1}{\text{Mcm}}$$

$$\frac{dA}{dt} = 0,002 \frac{1}{s}$$

$$l = 0,2 \text{ cm k\u00e4v\u00e9t\u00e1}$$

$$0,2 \cdot 25000 = 5000 \frac{1}{\text{M}}$$

Lambert-Beer

$$A = \epsilon \cdot l \cdot c$$

$\epsilon \rightarrow$ absorpci\u00f3s
koeffici\u00e9ns

$l \rightarrow$ re\u00e7eg vastags\u00e1g

$c \rightarrow$ konc.

$$A = \epsilon \cdot l \cdot c \quad \left| \frac{d}{dt} \right|$$

$$\frac{d}{dt}(A) = \frac{d}{dt}(\epsilon \cdot l \cdot c) = \epsilon \cdot l \cdot \frac{dc}{dt}$$

$\frac{dc}{dt} \rightarrow$ koncentraci\u00f3 id\u00f3
s\u00e9rt\u00e9si deriv\u00e1ltja
a reakci\u00f3 sebess\u00e9g

$$v = \frac{1}{\epsilon \cdot l} \cdot \frac{dA}{dt} = \frac{1}{5000 \frac{1}{\text{M}}} \cdot 0,002 \frac{1}{s}$$

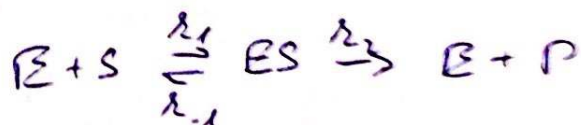
$$v = 4 \cdot 10^{-8} \frac{\text{M}}{s} = \underline{\underline{400 \text{ nM/s}}}$$

Michaelis-Menten modell

$$V_0 = \frac{V_{\max} [S]}{K_M + [S]}$$

Steady State *

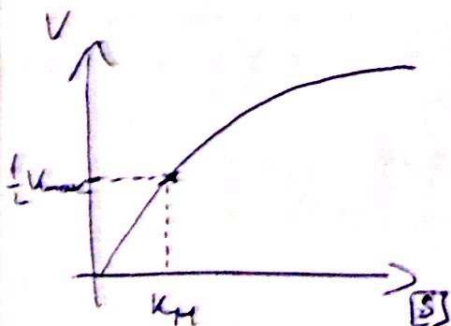
$$\Rightarrow \frac{d[ES]}{dt} = 0$$



$$K_M = \frac{k_{-1} + k_2}{k_1}$$

$$K_M = [S]$$

$$\rightarrow \frac{V_{\max} [S]}{[S] + [S]} = \frac{1}{2} V_{\max}$$



$$\frac{d[P]}{dt} = k_2 [ES] \quad \frac{d[ES]}{dt} = k_1 [E] [S] - (k_2 + k_{-1}) [ES] = 0 \quad *$$

Steady State $\rightarrow [ES]$ \u00e1lland\u00f3

k_{cat}/K_M , deviació

	$[S]$	$\frac{dA}{dt}$	V_0
1,	20 μM	0,00118 $\frac{1}{s}$	$4,72 \cdot 10^{-8}$
2,	40 μM	0,00223 $\frac{1}{s}$	$8,92 \cdot 10^{-8}$
3,	60 μM	0,00335 $\frac{1}{s}$	$3,34 \cdot 10^{-7}$

$$V_0 = \frac{V_{max} \cdot [S]}{K_M + [S]} \quad (*)$$

$$S \ll K_M *$$

$$(*) \frac{V_{max} [S]}{K_M}$$

$$V_0 = \frac{[S] \cdot k_{cat} \cdot [E]_T}{K_M} = \frac{k_{cat}}{K_M} \cdot \underbrace{[E]_T}_{\text{analitikus}} \cdot [S] \quad V_{max} = k_2 [E]_T, \quad k_2 = k_{cat}$$

hatekonyság

$$V_0 = \frac{dA/dt}{\epsilon \cdot l} = \frac{1}{\epsilon} \cdot \frac{dA}{dt}$$

$$\epsilon = 25000 \cdot \frac{1}{M \cdot cm}$$

1 $\mu g/ml$ MW = 25 kDa $\rightarrow$ enzim

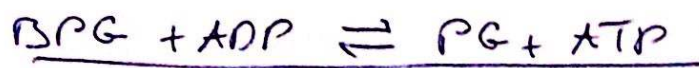
$$C_M = \frac{C_T}{M} = \frac{1 \mu g/ml}{25 kDa} = \frac{0,001 g/l}{25000 g/mol} = 4 \cdot 10^{-8} mol/l = [E]_T$$

$$\frac{k_{cat}}{K_M} = \frac{V_0}{[E]_T [S]} \Rightarrow$$

1,	59000
2,	55450
3,	55833
$\bar{x} = 56861$	

$$s = \sqrt{\frac{1}{2} (2139^2 + \dots)} = 185$$

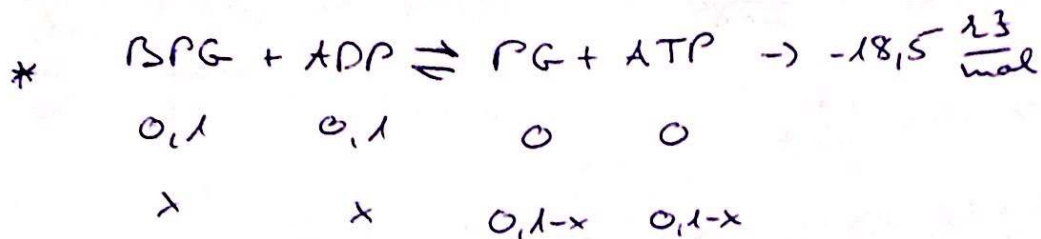
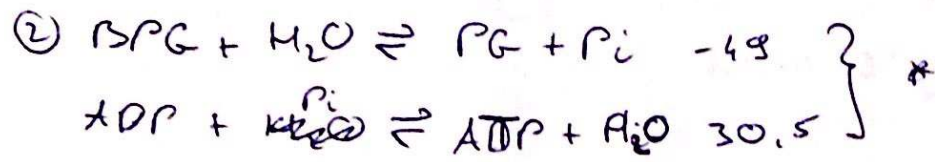
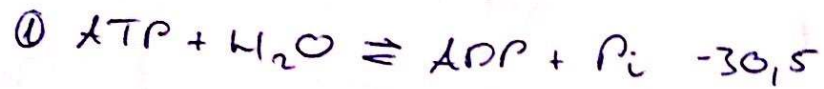
st. deviació: $5,69 \pm 0,19 \cdot 10^4$



$$R = 8,31 \text{ J/molK}$$

$$\textcircled{1} \text{ ATP} \rightarrow \Delta G^\circ = -30,5 \text{ kJ/mol}$$

$$\textcircled{2} \text{ BPG} \rightarrow \Delta G^\circ = -49 \text{ kJ/mol}$$



$$\Delta G = \Delta G^\circ + R \cdot T \cdot \ln \frac{[\text{PG}][\text{ATP}]}{[\text{ADP}][\text{BPG}]}$$

$$\Delta G^\circ = -RT \cdot \ln K \rightarrow \text{equilibrium constant}$$

$$-18500 \frac{\text{J}}{\text{mol}} = -8,31 \frac{\text{J}}{\text{molK}} \cdot 310\text{K} \cdot \ln \frac{[\text{PG}][\text{ATP}]}{[\text{ADP}][\text{BPG}]}$$

$$K = \frac{(0,1-x)^2}{x^2} = \frac{0,1^2 - 0,2x + x^2}{x^2}$$

$$K = e^{-\frac{\Delta G^\circ}{RT}} = e^{-\frac{18500}{8,31 \cdot 310}} = \underline{\underline{1315}}$$

$$1315 = \frac{(0,1-x)^2}{x^2}$$

$$36,26 = \frac{0,1-x}{x}$$

$$x = \frac{0,1}{37,26} = 0,00268$$

$$\text{riindular: } C = 2,68 \cdot 10^{-3} \text{ M}$$

$$\text{termel: } C = 0,09734 \text{ M}$$

Membrane pot.

$$F = 96485 \text{ C/mol}$$

$$\Delta U = -0,15 \text{ V}$$

$$R = 8,314 \text{ J/molK}$$

$$\text{ATP hidr.} \rightarrow \Delta G = -47,13 \text{ J/mol}$$

$$\frac{[H^+]_{\text{out}}}{[H^+]_{\text{in}}} = 26$$



$$\Delta G = R \cdot T \cdot \ln \frac{[H^+]_{\text{out}}}{[H^+]_{\text{in}}} + Z \cdot F \cdot \Delta U$$

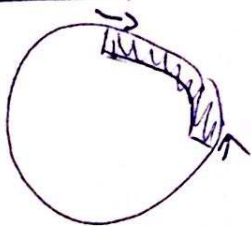
$$\Delta G = 22,87 \frac{\text{kJ}}{\text{mol}}$$

$$70\% \sim \underline{\underline{16,01 \frac{\text{kJ}}{\text{mol}}}}$$

$$\text{ATP hidr.} \rightarrow -47 \frac{\text{kJ}}{\text{mol}}$$

$$\frac{47 \text{ kJ/mol}}{16,01 \text{ kJ/mol}} = 2,93 \approx \underline{\underline{3 p^+}}$$

Plasmid



plasmid $\rightarrow$ 5000 bp

celgen $\rightarrow$ 1000 bp

$$80 \mu\text{g} = 80 \cdot 10^{-12} \text{ g plasmid}$$

$$\frac{80}{5} \cdot 10^{-12} \text{ g celgen} = 16 \cdot 10^{-12} \text{ g} \rightarrow 1 \mu\text{g cell!}$$
$$= 10^{-6} \text{ g}$$

$$16 \cdot 10^{-12} \text{ g} \cdot x = 10^{-6} \text{ g}$$

$$x = \frac{10^{-6}}{16 \cdot 10^{-12}} = 62500 \Rightarrow \text{empirische Zell}$$

Tudor, in 4 PCR ciklusban duplázódik

$$62500 = 2^x$$

$$x = \log_2 62500 = 15,93 \sim \underline{\underline{16 ciklus}}$$

PCR beads

① 26

② 33

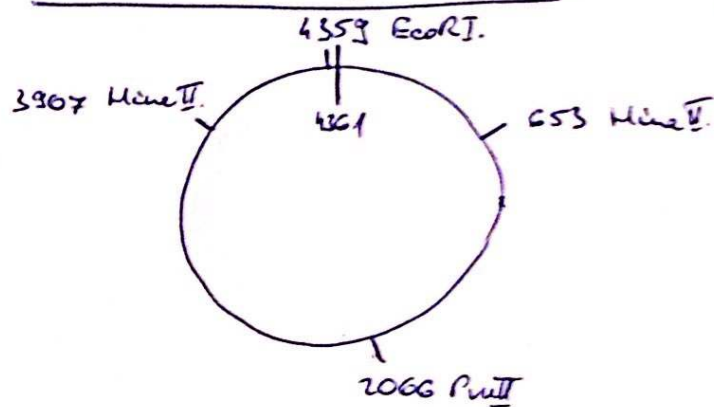
$$\Delta n = 7 \Rightarrow 2^7 = 128$$

$\Rightarrow$ 128 scan table } 100% - as
material usage

$$94\% - \text{ndel} \quad 100 + 94 \Rightarrow 1.94$$

$$1.94^7 = \underline{103.42 \text{ scans}}$$

Restriction's enzyme



4361 bp are given

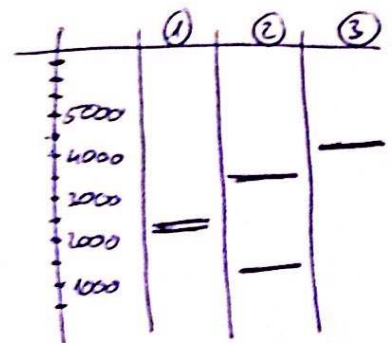
① PstI + EcoRI.
 $4359 - 2066 = 2293 \text{ bp}$

+ $4361 - 2293 = 2068 \text{ bp}$

② $3907 - 653 = 3254 \text{ bp}$
 $4361 - 3254 = 1107 \text{ bp}$



③ EcoRI.
 $4361 - 0$



Membrane diffusion

$$\lambda = (4 \cdot D \cdot t)^{1/2}$$

$$D = 1 \mu\text{m}^2/\text{s}$$

$$d = 3 \mu\text{m} \Rightarrow r = 1.5 \mu\text{m}$$

$$\lambda = \frac{3}{2} \pi \cdot \frac{1}{2} \cdot 2 = \frac{3}{2} \pi$$

$$\frac{3}{2} \pi = \sqrt{1 \cdot t \cdot 4}$$

$$\frac{3}{2} \pi = 2 \sqrt{t}$$

$$t = \frac{9}{16} \pi^2 \text{ s}$$

$$K = \frac{2\pi r}{2\pi r}$$